

Un Steady state Conduction [Transient Conduction]

→ when the heat energy is being added & removed to & from a body, its energy content (Internal energy) changes, resulting into change in its temperature at each point within the body over the time.

→ In unsteady state conduction Temperature varies with time

$$\frac{dT}{dt} \neq 0.$$

→ During this ~~period~~ transient period, the temperature becomes function of time as well as direction in the body. The conduction occurred ~~thru~~ during this period is called Transient conduction.

→ The temperature and Heat ~~conduction~~ are depends both on the time and space coordinates, i.e. $T = f(x, y, z, t)$

→ In many engineering applications, the heat transferred is transient. The heat treatment process like Quenching, Annealing, normalizing etc... are process of unsteady state

→ change in temperature during unsteady state may follow a periodic (d) a non-periodic variation.

→ The temperature changes in repeated cycles and the conditions get repeated after some fixed time interval is called periodic variation.

→ Temperature variations in the cylinders of ~~an~~ an I-C engine are considered periodic with respect to ~~the~~ crank angle

→ ~~Ex1~~ periodic heat flow in building between day and night.

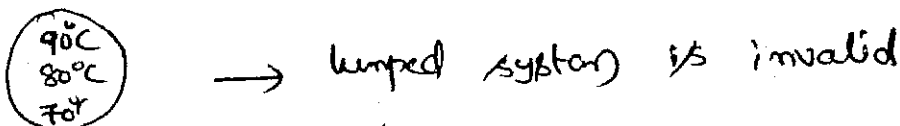
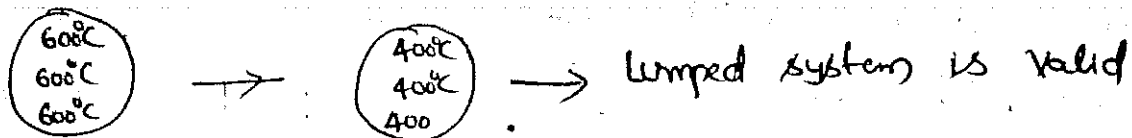
→ ~~[contributing to discomfort]~~
The temperature changes as some non-linear function of time. This variation is neither according to definite pattern nor is repeated cycles. It is called non-periodic variation.

Ex1- heating of an ingot in furnace, cooling of balls

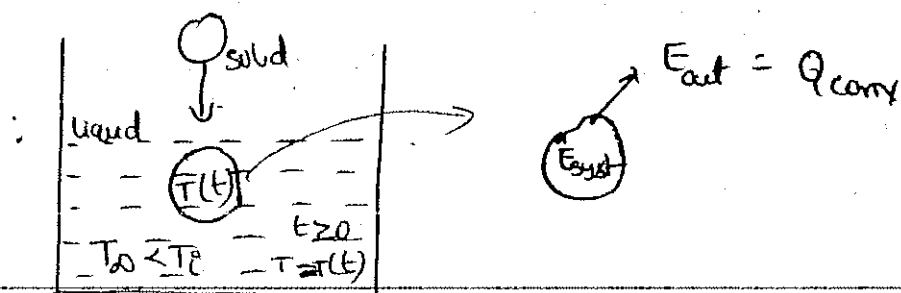
Lumped system Analysis :- ①

In Lumped system Analysis temperature varies with time, but not with space i.e., a lumped system analysis temp is the function of time only.

Ex1:- A Copper ball taken from oven



Equation for time taken by an object to reach a particular temperature :-



solid suddenly exposed to convection environment T_∞ .

Let T be Temp of the object at any temperature ' t '

②

T_i be the temp of initial temp of the object

A be the surface area of the object

h be the heat transfer coefficient

Heat convected from the surface = Energy stored in the object

$$\text{Heat convected} = hA(T_0 - T)dt$$

$$\text{Energy stored} = mc_p dT$$

$\rightarrow m = \text{mass of object}$

$c_p = \text{sp. heat of the object}$

$$\therefore hA(T_0 - T)dt = mc_p dT$$

$$\rho = \frac{m}{V}$$

$$hA(T_0 - T)dt = \rho V c_p dT$$

$$m = \rho V$$

$$hA(T_0 - T)dt = \rho c_p V \cdot d(T)$$

$$-hA(T - T_0)dt = \rho c_p V \cdot d(T)$$

$$\int_0^t -\frac{hA}{\rho c_p V} \cdot dt = \int_{T_i}^T \frac{d(T)}{(T - T_0)}$$

$$-\frac{hA}{\rho c_p V} \int_0^t dt = \int_{T_i}^T \frac{d(T)}{T - T_0}$$

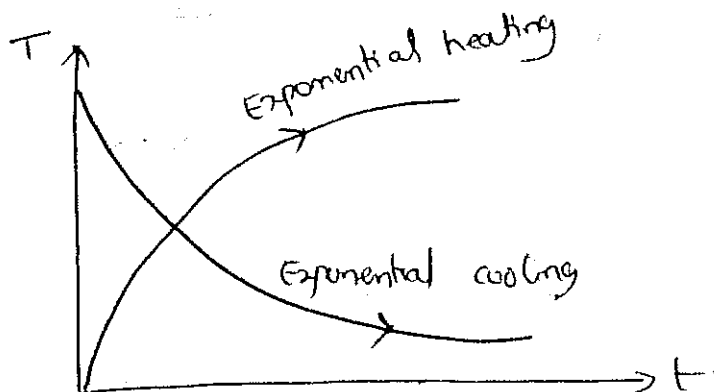
$$-\frac{hA}{\rho c_p V} [t]_0^t = \ln(T - T_0) \Big|_{T_i}^T$$

$$\frac{-hA}{\rho c V} t = \ln \left[\frac{T - T_{\infty}}{T_i - T_{\infty}} \right] \quad (2)$$

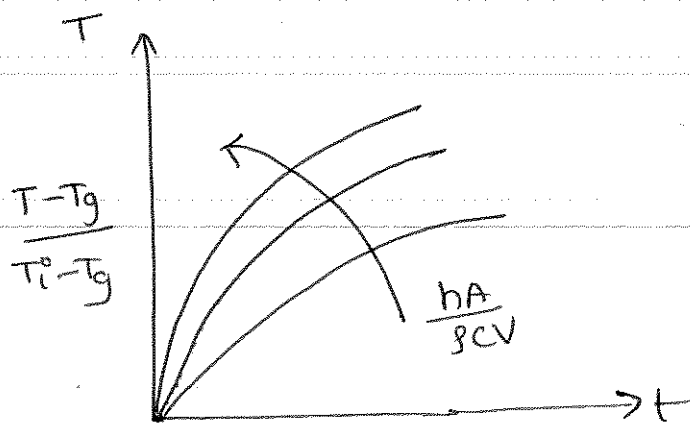
$$\frac{T - T_{\infty}}{T_i - T_{\infty}} = e^{\frac{-hA}{\rho c V} t}$$

OBSERVATION OF ABOVE EQUATION:-

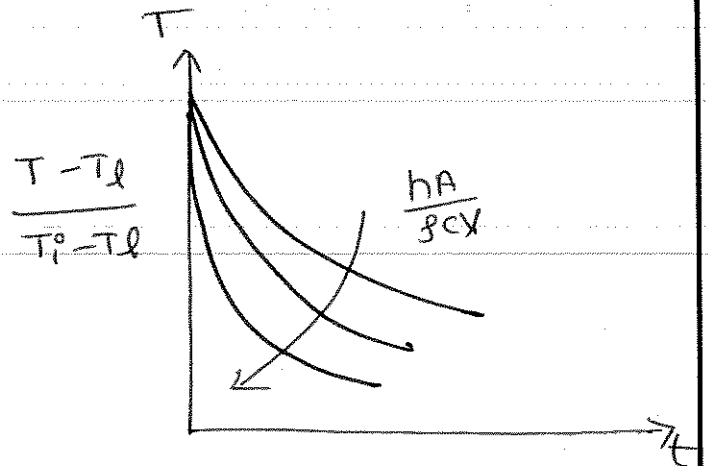
- 1) The equation can be used to determine the time t required for solid to reach some temperature T
- 2) The temp of a body approaches the ambient temp T_{∞} exponentially. The temp of the body changes rapidly at the beginning [due to large temp difference], but slows down later on.
- 3) The large value of $\frac{hA}{\rho c V}$ indicates that the body will approach the ambient temp in a short time
- 4) The temperature varies with time exponentially.



Heating and cooling of body when $k \rightarrow \infty$



T_g - gas temp $>$ Body temp



T_l - liquid temp $<$ Body temp

→ ~~For~~ ~~Variable~~

Biot Number :-

→ In general accepted that lumped system analysis is applicable if

$$Bi \leq 0.1$$

→ when $Bi \leq 0.1$, the variation of temp with location within the body is slight and can reasonably be approximated as being uniform.

→ The first step in the application of lumped system analysis is the calculation of Biot number.

→ If the Biot number (Bi) is less than 0.1 we should approach lumped system analysis, otherwise we should follow graphical method of ~~Heisler~~ Heisler charts and Crober charts

→ Biot number is the ratio of the conductive resistance to the convective resistance

$$\frac{R_{\text{cond}}}{R_{\text{conv}}} = \frac{\frac{L}{KA}}{\frac{1}{hA}} = \frac{L}{KA} \times \frac{hA}{1} = \frac{hL}{K}$$

(OR)

→ Ratio of the convection at the surface to conduction within the body.

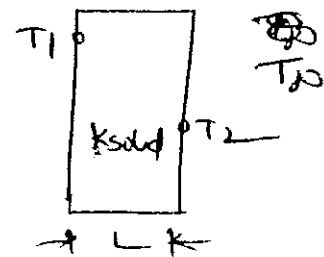
$$\frac{Q_{\text{conv}}}{Q_{\text{cond}}} = \frac{hA \, dT}{KA \frac{dT}{dx}} = \frac{h \, dx}{K} = \frac{hL}{K}$$

(OR)

→ It is the ratio of temp drop due to conduction to the temp drop due to convection

$$\frac{KA(T_1 - T_2)}{L} = hA(T_2 - T_\infty)$$

$$\frac{T_1 - T_2}{T_2 - T_\infty} = \frac{hAL}{KA} = \frac{\frac{L}{KA}}{\frac{1}{hA}}$$



$$= \frac{\text{conductive resistance}}{\text{convective resistance}} = Bi^0$$

$$\therefore \frac{T_1 - T_2}{T_2 - T_\infty} = \frac{hL}{K_{\text{solid}}} = Bi^0$$

$$\Rightarrow Bi^0 = \frac{T_1 - T_2}{T_2 - T_\infty} = \frac{hL}{K_{\text{solid}}} = \frac{\Delta T_{\text{cond}}}{\Delta T_{\text{conv}}} = \frac{R_{\text{cond}}}{R_{\text{conv}}}$$

$$Bi^0 = \frac{ICR}{ECR}$$

ICR $\rightarrow$ Internal conductive resistance
 L/kA

ECR $\rightarrow$ External conductive resistance

$$ICR = \frac{L}{kA}$$

$$\uparrow \frac{L}{kA} = ICR \downarrow = \text{Bi}^0 \downarrow$$

$$Bi^0 \downarrow \frac{ICR}{ECR} = Bi^0 \downarrow \quad \& \quad \uparrow \frac{ICR}{ECR} = Bi^0 \uparrow$$

- $\rightarrow$ The Biot number should be as small as possible for lumped system analysis to be applicable.
- $\rightarrow$ Small bodies with high thermal conductivity are good condition for lumped system analysis, especially when they are in a medium that is poor conductor of heat (such as air (or) another gas) and motion less.
- $\rightarrow$ Lumped system is valid for $Bi^0 < 0.1$.

$$Bi^0 = \frac{hL_c}{k_{solid}}$$

$L_c = \text{characteristic length}$

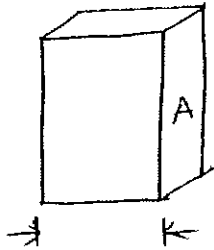
$$L_c = \frac{\text{Volume of body}}{\text{surface area of body}} = \frac{V}{A}$$

Characteristic lengths L_c for different shapes:-

1) plane slab with convection at both sides

$$V = AL$$

$$A_s = A + A = 2A$$



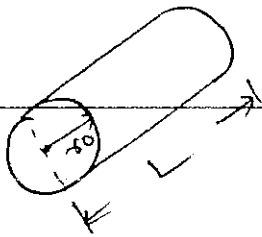
$$L_c = \frac{V}{A_s} = \frac{AL}{2A} = L/2$$

$$\therefore L_c = L/2$$

2) For solid cylinder of radius r_0 and length L (short cylinder)
Both sides convection

$$V = \pi r_0^2 L$$

$$A = 2\pi r_0^2 + 2\pi r_0 L$$



$$L_c = \frac{\pi r_0^2 L}{2\pi r_0^2 + 2\pi r_0 L} = \frac{\pi r_0^2 L}{2\pi r_0(r_0 + L)}$$

$$L_c = \frac{r_0 L}{2(r_0 + L)}$$

$$L_c = \frac{\pi r_0^2 L}{2\pi r_0^2 + 2\pi r_0 L}$$

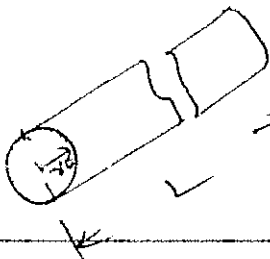
$$= \frac{\pi r_0^2 L}{2\pi r_0(r_0 + L)}$$

$$\therefore L_c = \frac{r_0 L}{2(r_0 + L)}$$

3) For Long cylinder of radius r_0 ($L \gg r_0$)

$$V = \pi r_0^2 L$$

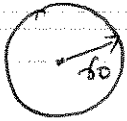
$$A = 2\pi r_0 L$$



$$L_c = \frac{\pi r_0^2 L}{2\pi r_0 L} = \frac{r_0}{2} = \frac{D}{4}$$

$$\therefore L_c = \frac{r_0}{2} = \frac{D}{4}$$

4) For solid sphere of radius r_0



$$V = \frac{4}{3} \pi r_0^3$$

$$A = 4 \pi r_0^2$$

$$L_c = \frac{\frac{A}{3} \pi r_0^3}{4 \pi r_0^2} = \frac{r_0}{3}$$

$$L_c = \frac{r_0}{3} = \frac{D}{6}$$

5) For a cube of a side L

$$V = L^3$$

$$A = 6L^2$$

$$L_c = \frac{L^3}{6L^2} = \frac{L}{6}$$

$$L_c = \frac{L}{6}$$

Fourier Number :-

→ The degree of penetration of heating (or) cooling effect through the solid is called Fourier number $[Fo]$.

$$\begin{aligned} \rightarrow \frac{hA}{\rho c V} t &= \frac{h}{\rho c L_c} t = \frac{h t}{\rho c L_c} \times \frac{L_c}{L_c} \times \frac{K}{K} \quad \left| \frac{h L_c}{K} \right. \\ &= Bi^* \cdot \frac{K}{\rho c} \cdot \frac{t}{L_c^2} \\ &= Bi^* \alpha \cdot \frac{t}{L_c^2} \\ \left[\frac{hA}{\rho c V} t \right] &= Bi^* \cdot Fo \end{aligned}$$

$$\therefore \text{Fourier NO} = \frac{\alpha t}{L_c^2}$$

→ It is also defined as the ratio of the rate of heat conduction to the ~~the~~ thermal energy storage in the solid.

conventional questions:-

- i) ~~the~~ In a quench hardening process, steel rods ($\rho = 7832 \text{ kg/m}^3$, $c_p = 434 \text{ J/kgK}$, $k = 63.9 \text{ W/mK}$) are heated in a furnace to 850°C and then cooled in a water bath to an average temp of 95°C . The water bath has a uniform temperature of 40°C and convection heat transfer coefficient of $450 \text{ W/m}^2\text{K}$. If the steel rod have a diameter of 50 mm and a length of 2 m , determine
 - i) Time required to cool the steel rod from 850°C to 95°C in the water bath
 - ii) The total amount of heat transferred to water during the quenching of a single rod

Data

$$\rho = 7832 \text{ kg/m}^3, \quad c_p = 434 \text{ J/kgK}, \quad k = 63.9 \text{ W/mK}$$

$$T_i = 850^\circ\text{C}, \quad T_{\infty} = 95^\circ\text{C}, \quad T_b = 40^\circ\text{C}$$

$$h = 450 \text{ W/m}^2\text{K}$$

$$D = 50 \text{ mm} = 0.05 \text{ m}, \quad L = 2 \text{ m}$$

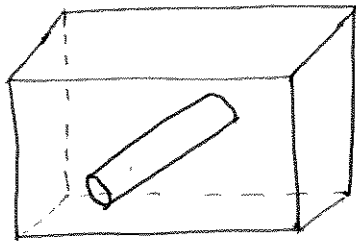
$$R = 0.025 \text{ m}$$

Find:-

- 1) $t = ?$ to reach 850°C to 95°C
- 2) Total Heat transferred during cooling

Assumptions :-

- 1) Thermal properties of steel rod are constant
- 2) convection heat transfer coefficient is uniform
- 3) Heat transfer by radiation is negligible

Schematic :-Solution:-

For proceeding the problem first we should find out Biot number (Bi - non dimensional no.) & if the Bi is less than 0.1 we can follow lumped capacitance method otherwise we should follow Heisler ~~charts~~ ~~and~~ ~~charts~~ charts

$$Bi = \frac{h L_c}{k}$$

$$L_c = \frac{V}{A_s} = \frac{\frac{\pi D^2}{4} L}{\pi D L} = \frac{D}{4} = \frac{0.05}{4} = 0.0125 \text{ m}$$

$$Bi = \frac{450 \times 0.0125}{63.9} = 0.88$$

$\therefore Bi > 0.1$, $\therefore$ then we follow Lumped system Analysis

$$\frac{T - T_\infty}{T_i - T_\infty} = e^{-\left(\frac{h A}{\rho c_p V}\right) t} = e^{-\left(\frac{h}{\rho c L_c}\right) t}$$

$$\frac{95-40}{850-40} = e^{-\left(\frac{450}{7832 \times 434 \times 0.0125}\right) t}$$

$$\ln\left(\frac{95-40}{850-40}\right) = -\left(\frac{450}{7832 \times 434 \times 0.0125}\right) \times t$$

$$\therefore t = 254 \text{ sec.}$$

2)

$$Q = m c_p (T_i - T_o)$$

$$\rho = \frac{m}{V}$$

$$m = \rho \times V$$

$$= \rho V c_p (T_i - T_o)$$

$$= \frac{\rho \pi D^2 L c_p (T_i - T_o)}{4}$$

$$= \frac{7832 \times \pi \times 0.05^2 \times 2 \times 434 \times (850 - 40)}{4}$$

$$\therefore Q = \underline{\underline{10.1 \text{ MJ}}}$$

2) An egg with mean diameter of 4cm is initially at 25°C. It is placed in boiling water for 4min and found to be ~~consume~~ consumers taste. For how long should be a ~~small quantity~~ similar egg for same consumer to be boiled when taken from refrigerator at 2°C. Use lumped system analysis and take the thermophysical properties of egg as.

$$k = 12 \text{ W/mK}, \quad h = 125 \text{ W/m}^2\text{K}, \quad c = 2000 \text{ J/kgK}$$

$$\rho = 1250 \text{ kg/m}^3$$

data

case 1:-

$$D = 4 \text{ cm} = 0.04 \text{ m}$$

$$r_o = 0.02 \text{ m}, \quad T_i = 25^\circ\text{C}, \quad T_o = 100^\circ\text{C}$$

$$\text{time}(t) = 4 \text{ min} = 240 \text{ sec}$$

Case 2:- Similar egg (geometric properties are same)

$$D = 4\text{cm} = 0.04\text{m}$$

$$B_0 = 0.02\text{m}$$

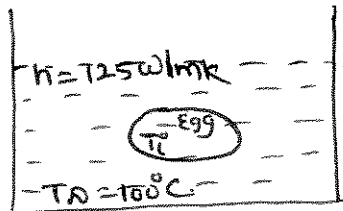
$$T_i = 2^\circ\text{C}$$

$$k = 12\text{W/mK}, h = 125\text{W/m}^2\text{K}, c = 2000\text{J/kgK}, \rho = 1250\text{kg/m}^3$$

Find:-

- 1) Time required for egg at 2°C to be boiled to consumers taste.

schematic:-



Assumptions:-

- 1) Egg as a sphere
- 2) constant properties
- 3) Boiling temp of water as 100°C

Sol:-

For proceeding the problem first we should find out B_i , then proceed further processes

$$B_i = \frac{hL_c}{k}$$

$$L_c = \frac{V}{A_s}$$

$$= \frac{R}{3} = \frac{0.02}{3} = 0.00666\text{m}$$

$$B_i = \frac{125 \times 0.00666}{12}$$

$$B_i = 0.06944$$

$$Bi < 0.1$$

$\therefore$ Bi number is less than 0.1 then lumped system Analysis is applicable

$$\frac{T - T_{\infty}}{T_i - T_{\infty}} = e^{-\left(\frac{hA}{\rho c V}\right)t}$$

$$\frac{T - T_{\infty}}{T_i - T_{\infty}} = e^{-\left(\frac{h\rho}{\rho c L_c}\right)t}$$

$$\frac{T - 100}{25 - 100} = e^{-\left(\frac{125}{1250 \times 2000 \times 0.02}\right)t}$$

$$T = 87.6^{\circ}\text{C}$$

When egg taken ~~from~~ from refrigerator

$$T_i = 2^{\circ}\text{C}, \quad T = 87.6^{\circ}\text{C}$$

$$\frac{T - T_{\infty}}{T_i - T_{\infty}} = e^{-\left(\frac{h}{\rho c L_c}\right)t}$$

$$\ln \left[\frac{87.6 - 100}{2 - 100} \right] = e^{-\left(\frac{125}{1250 \times 2000 \times 0.02}\right)t}$$

$$\therefore t = 275.6 \text{ sec.}$$

$$t = 4.6 \text{ min}$$

3) A person is found dead at 5 PM in a room, whose temp is ~~20~~ 25°C . The temperature of the body is measured to be 25°C , when found and heat transfer coefficient is estimated to be $8 \text{ W/m}^2\text{K}$. Modelling the human body as 30 cm diameter, 1.7 m long cylinder, calculate actual time

of death of person. take Thermophysical properties of body.

$$K = 6.08 \text{ W/mK}, \rho = 900 \text{ kg/m}^3, c = 4000 \text{ J/kgK}.$$

Data:-

$$T_D = 20^\circ\text{C}$$

$$K = 6.08 \text{ W/mK}$$

$$T = 25^\circ\text{C}$$

$$\rho = 900 \text{ kg/m}^3$$

$$h = 8 \text{ W/m}^2\text{K}$$

$$c = 4000 \text{ J/kgK}$$

$$D = 30 \text{ cm} = 0.3 \text{ m}$$

$$r_0 = 0.15 \text{ m}, L = 1.7 \text{ m}$$

Find:-

- 1) Actual time of death person

Assumptions:-

- 1) Healthy person, the body temp of 37°C at the time of death.
- 2) NO Radiation heat transfer
- 3) Constant properties

Sol:-

First find out Bi^0

Body treated as a short cylinder

$$Bi^0 = \frac{hL_c}{K}$$

$$L_c = \frac{V}{A_s} = \frac{\pi r_0 L}{2\pi r_0 L + 2\pi r_0^2}$$

$$L_c = \frac{r_0 L}{2(r_0 + L)} = \frac{0.15 \times 1.7}{2(0.15 + 1.7)}$$

$$L_c = 0.0689 \text{ m}$$

$$\therefore Bi^0 = \frac{8 \times 0.0689}{6.08} = 0.092$$

$\therefore Bi^0 < 0.1$, so lumped system is applicable

$$\frac{T - T_\infty}{T_i - T_\infty} = e^{-\left(\frac{hA}{\rho c_p V}\right)t}$$

$$\frac{T - T_\infty}{T_i - T_\infty} = e^{-\left(\frac{h}{\rho c_p L}\right)t}$$

$$\ln \left[\frac{25 - 20}{37 - 20} \right] = -\left(\frac{8}{900 \times 4000 \times 0.0689} \right)t$$

$$t = 37,943 \text{ sec.}$$

$$t = 10.54 \text{ hours before}$$

4) In a quenching process, a copper plate 3mm thick is heated upto 400°C and then exposed to an ambient at 25°C . with the convection co-efficient of $28 \text{ W/m}^2\text{K}$ calculate the time required for the plate to reach temp of 50°C . Take thermophysical properties as $c = 380 \text{ J/kgK}$, $\rho = 880 \text{ kg/m}^3$, $k = 38 \text{ W/mK}$.

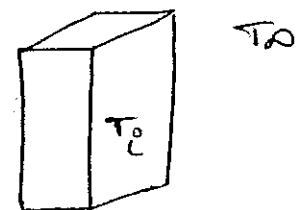
Data

$$L = 3 \text{ mm} = 3 \times 10^{-3} \text{ m}$$

$$T_i = 400^\circ\text{C}, T_\infty = 25^\circ\text{C}$$

$$h = 28 \text{ W/m}^2\text{K}, T = 50^\circ\text{C}$$

$$c = 380 \text{ J/kg}, \rho = 880 \text{ kg/m}^3, k = 38 \text{ W/mK}$$



Find :-

1) Time required to reach temp from 400 to 50°C .

Sol

For proceeding the problem we should find out Bi number, if the obtained Bi is < 0.1 we should follow

Follow the lumped system Analysis otherwise follow Heisler charts ~~for~~ ~~find~~

$$Bi^0 = \frac{hL_c}{k},$$

$$L_c = \frac{V}{A_s} = \frac{L}{2} = 0.0015 \text{ m}$$

$$Bi^0 = \frac{28 \times 0.0015}{38} \Rightarrow Bi^0 = 1.0909 \times 10^{-4}$$

$\therefore Bi^0 \leq 0.1 \rightarrow$ use lumped system

$$\frac{T - T_\infty}{T_i - T_\infty} = e^{-\left(\frac{h}{\rho c L_c}\right) t}$$

$$\ln \left[\frac{50 - 25}{400 - 25} \right] = - \left(\frac{28}{880 \times 380 \times 0.0015} \right) \times t$$

$$\therefore t = 485 \text{ sec} = \underline{\underline{8.08 \text{ min}}}$$

- 5) A Titanium alloy blade of an axial compressor for which $k = 25 \text{ W/m}^2\text{K}$, $\rho = 4500 \text{ kg/m}^3$, $c = 520 \text{ J/kgK}$ is initially at 60°C . The effective thickness of the blade is 10 mm and then exposed to gas stream at 600°C . The blade experiences a heat transfer coefficient of $500 \text{ W/m}^2\text{K}$, use low Biot number approximation to estimate temp of blade after 1, 5, 20 & 100 sec.

Data $k = 25 \text{ W/m}^2\text{K}$, $\rho = 4500 \text{ kg/m}^3$, $c = 520 \text{ J/kgK}$
 $T_i = 60^\circ\text{C}$, $L = 10 \text{ mm} = 10 \times 10^{-3} \text{ m}$, $T_\infty = 600^\circ\text{C}$
 $h = 500 \text{ W/m}^2\text{K}$

Find:-

1) Temperatures at 1, 5, 20 & 100 sec

Sol

$$Bi = \frac{h r_c}{k} \quad L_c = \frac{10 \times 10^{-3}}{2} = 5 \times 10^{-3} \text{ m}$$

∴ Low Bi number approximation is

$$\frac{T - T_\infty}{T_i - T_\infty} = e^{-\left(\frac{h}{\rho c L_c}\right) t}$$

$$\frac{T - 600}{60 - 600} = e^{-\left(\frac{500}{4500 \times 520 \times 5 \times 10^{-3}}\right) \times 1}$$

$$T = 82.6^\circ\text{C}$$

and Further do same procedure for 5, 20, 100 sec

- 6) A Thermocouple junction, which may be approximated as a sphere; It is to be used for temp measurement in a gas stream. The convection b/w the junction surface and the gas $h = 400 \text{ W/m}^2\text{K}$ and junction Thermophysical properties are $k = 20 \text{ W/mK}$, $c = 400 \text{ J/kgK}$, $\rho = 8500 \text{ kg/m}^3$. Determine the junction diameter needed for the thermocouples to have a time constant of 1 sec. If the thermocouple junction is at 25°C and is placed in a gas stream that is at 200°C , how long it will take for the junction to reach 199°C .

data

$$h = 400 \text{ W/m}^2\text{K}, k = 20 \text{ W/mK}, c = 400 \text{ J/kgK}, \rho = 8500 \text{ kg/m}^3$$

$$T_f = 25^\circ\text{C}, T_\infty = 200^\circ\text{C}, T = 199^\circ\text{C}$$

Find

1) Dia of Thermocouple (d) = ?

2) Time required to reach temp of Film from 25°C to 199°C .Sol

According to time response and response of Thermocouples

$$t = \frac{\rho V c}{h A_s}$$

$$t = \frac{\rho c b c}{h}$$

$$1 = \frac{8500 \times 400 \times L_c}{400}$$

$$\Rightarrow L_c = 1.176 \times 10^{-4} \text{ m}$$

$$\therefore L_c = \frac{V}{A} \quad \text{Thermocouples is a sphere}$$

$$\therefore \text{for sphere } L_c = \frac{D}{6} \Rightarrow D = 7.058 \times 10^{-4} \text{ m}$$

 $\therefore$ further proceeding is

$$Bi = \frac{h L_c}{k} \Rightarrow Bi = \frac{400 \times 7.058 \times 10^{-4}}{20} = 0.014$$

 $\therefore Bi \leq 0.1 \rightarrow$ lumped system analysis should follow

$$\eta \left| \frac{T - T_\infty}{T_f - T_\infty} \right| = e^{-\left(\frac{h}{\rho c L_c}\right) t} \Rightarrow \left| \frac{199 - 200}{25 - 200} \right| = e^{-\frac{400}{8500 \times 400 \times 7.058 \times 10^{-4}} t}$$

$$\therefore t = \underline{\underline{5.2 \text{ sec}}}$$

Time Constant and Response of a Thermocouple :-

- When two dissimilar metals are joined together at two points to form a couple loop and a temperature difference exists b/w the junctions, an electrical potential is setup b/w the junction.
- Such Arrangement is known as Thermocouple and used for the measuring of temperature.
- measurement of temperature by thermocouple is an important application of the lumped system analysis.
- The ~~Response~~ ["]Response of a thermocouple["] is defined as the time required for the thermocouple to reach the source temperature when it is exposed to it.
- The Time constant Thermocouple is determined from lumped parameters & it is finally

$$\frac{T - T_0}{T_i - T_0} = e^{-\left(\frac{hA}{\rho V c}\right)t}$$

- The sensitivity of the Thermocouple is defined as the time required by the thermocouple to reach 63.2% of its steady state value.

$$\frac{T - T_0}{T_i - T_0} = e^{-\left(\frac{hA}{\rho V c}\right)t} = 1 - 0.368 = 0.632$$

$$1 - 0.632 = e^{-\left(\frac{hA}{\rho V c}\right)t}$$

$$0.368 = e^{-\left(\frac{hA}{\rho V c}\right)t}$$

$$e^{-\left(\frac{hA}{\rho V c}\right)t} = 0.368$$

$$1 - 0.632 = e^{-\left(\frac{hA}{\rho c v}\right) t}$$

$$0.368 = e^{-\left(\frac{hA}{\rho c v}\right) t}$$

$$e^{(-1)} = e^{-\left(\frac{hA}{\rho c v}\right) t} \quad \left| \quad e^{(-1)} = 0.368 \right.$$

$$1 = e^{\frac{hA}{\rho c v} t}$$

$$\therefore t = \frac{\rho v c}{hA} = \frac{\rho c}{h L_c}$$

→ The time constant is indicative of the speed of response
 i.e. 1) $\uparrow t \Rightarrow$ slow of the response time
 2) $\downarrow t \Rightarrow$ high of the response time

→ so as we are engineers to develop a thermocouple to response fastly ~~to~~ and required low value of time & it can be achieved by.

- 1) Decreasing the diameter of wire
- 2) Using light materials of low density and low specific heat
 [Mean less Heat storage capacity]
- 3) Increasing the heat transfer coefficient.

→ Depending upon the type of fluid used, the response times for different sizes and materials of Thermocouple wires usually lie b/w. 0.04 to 2.5 seconds

Transient Temperature Charts :- [HEISLER & GROBER CHARTS]

- one the Bi number is known, ~~and~~ ~~the~~ ~~or~~ if we got low Bi numbrs we should not use lumped capacitance to find out time (t) required temp of Heat etc - -
- ~~For~~ when the internal temperature gradients are large, lumped heat capacitance system becomes unsuitable for the Analysis of Transient heat conduction problems.
- ~~For~~ In such situation the Heisler and grober charts are widely used
- The Transient temperature charts for a large plane wall, Long cylinder and sphere were represented by M.P. Heisler in 1947 and are called Heisler charts. They were supplemented in 1961 with heat transfer charts by H. Grober.
- There are Three charts associated with each Geometry.
- ① 1) The First chart is used for ^{to} determine ~~the~~ ~~the~~ temperature T_0 at the centre of the geometry at given time t .
- 2) The second chart is used for to determine the temperature at other locations at the same time in terms of T_0 .
- 3) The Third chart is to determine the total amount of heat transfer up to the time t .

→ Therefore The Heisler Charts are widely used for determine

- 1) centre line temperature
- 2) position temperature
- 3) The heat transfer

→ To obtain the desired value of unknowns, the various dimensionless parameters required are

Temperature

- 1) Biot number $[Bi^*]$
- 2) Fourier number $[Fo]$
- 3) Temperature ratio at the centre $[Q/Q_i]$
- 4) Temperature ratio at any position $[Q/Q_c]$
- 5) Dimensionless position $[X/L]$
- 6) Dimensionless Heat transfer $[Q/Q_i]$

1) Transient Temperature charts for slab :-

→ Consider a slab (i.e. plane) of thickness $2L$, confined to the region $-L \leq x \leq L$. The slab initially at temp T_i , is suddenly exposed to convection environment (for $t > 0$) with a heat transfer coefficient h , on its both boundary surfaces.

→ The heat ~~to~~ flow from both surfaces in-ward, Due to symmetry of problems, only half region $0 \leq x \leq L$ is considered

→ The dimensionless parameters for slab can be expressed as

1) Biot number $[Bi^*] = \frac{hL}{k}$

2) Fourier number $[Fo] = \frac{\alpha t}{L^2}$

3) Temperature ratio at the centre $\left[\frac{\theta_c}{\theta_i} \right] = \frac{T_0 - T_\infty}{T_i - T_\infty}$

4) Temperature ratio at any position $\left[\frac{\theta}{\theta_c} \right] = \frac{T(x,t) - T_\infty}{T_0 - T_\infty}$

5) Dimensionless position $= \frac{x}{L}$

6) Dimensionless heat Transfer $= \frac{Q}{Q_i}$

where

L = half thickness of a slab - m

x = position in the slab measured from centre where the temp is required - m

T_0 = centre line temp in the slab - $^{\circ}\text{C}$

$T(x,t)$ = position temp in the slab - $^{\circ}\text{C}$

T_i = Initial temp in the slab - $^{\circ}\text{C}$

t = time - seconds

Q = total amount of energy lost by plate during time t - J

Q_i = initial internal energy content in the slab

$$= m c_p \Delta T$$

$$= \rho V c_p \Delta T$$

$$= \rho (A 2L) c_p (T_i - T_\infty) \text{ - Joules}$$

→ The temp at any position x from the Mid-plane can be obtained from position correction temperature chart

$$\frac{\theta}{\theta_i} = \frac{\theta_c}{\theta_i} \times \frac{\theta}{\theta_c} = \frac{T - T_\infty}{T_i - T_\infty}$$

2) ~~Temperature Transient~~ ^{for} Charts

2) Transient Temperature charts for long cylinder & sphere:-

→ consider a long cylinder or a sphere of radius r_0 , initially at temp T_i is suddenly subjected to convection environment for ($t > 0$) with heat transfrs coefficient h and fluid temp T_∞ , is

1) Biot number $[Bi] = \frac{hr_0}{k}$

2) Fourier number $[Fo] = \frac{\alpha t}{r_0^2}$

3) Temp ratio at the centre $\frac{\theta}{\theta_c} = \frac{T_0 - T_\infty}{T_i - T_\infty}$

4) Temp ratio at any position $\frac{\theta}{\theta_c} = \frac{T(r,t) - T_\infty}{T_i - T_\infty}$

5) Dimensionless radial position $= \frac{r}{r_0}$

6) Dimensionless heat transfer $= \frac{Q}{Q_i}$

where

r_0 = radius of cylinder (or) sphere - m

r = position radius in cylinder (or) sphere - m

$T(r,t)$ = position temp - °C

α = Thermal diffusivity - m^2/s

k = Thermal conductivity = W/mK

h = heat transfrs coefficient = W/m^2K

Conventional questions :-

- 1) A 50mm thick iron plate is initially at 225°C . It is both surfaces are suddenly exposed to an environment at 25°C with convection co-efficient of $500\text{W/m}^2\text{K}$. Calculate
- 1) The centre temperature, 2 minutes after the state of exposure
 - 2) The temp at the depth of 10mm from the surface after 2min of exposure
 - 3) Energy removed from the plate per square meter during this period. Take Thermophysical properties of $k = 60\text{W/mK}$, $\rho = 7850\text{kg/m}^3$, $c = 460\text{J/kgK}$
 $\alpha = 1.6 \times 10^{-5}\text{m}^2/\text{s}$.

Data

$$2L = 50\text{mm} \Rightarrow L = 25\text{mm} = 0.025\text{m}$$

$$T_i = 225^{\circ}\text{C}, T_{\infty} = 25^{\circ}\text{C}, h = 500\text{W/m}^2\text{K}$$

$$t = 2\text{min} = 120\text{sec.}$$

Find :-

- 1) $T_{\text{centre}} = ?$
- 2) $T_{x/L} = ?$
- 3) $Q = ?$

Assumption :-

Sol :-

Considering the plate thickness $2L$, hence consider L_c as characteristic length ($2L/2$)

For doing transient problems, first we should find out Bi^o , if Bi^o is less than 0.1 follow lumped capacitance otherwise use Heisler charts

$$Bi^o = \frac{hL_c}{K} = \frac{500 \times 0.025}{60} = 0.21$$

$\therefore Bi^o > 0.1$ use Heisler charts for further proceedings

$$Fo = \frac{\alpha t}{L_c^2} = \frac{1.6 \times 10^{-5} \times 120}{(0.025)^2} = 3.07.$$

1) From Heisler chart of plane slab for centre line temperature from data book

$$\frac{\theta_c}{\theta_i^o} = \frac{T_0 - T_\infty}{T_i^o - T_\infty}$$

for $Bi^o = 0.21$ & $Fo = 3.07$ The $\frac{\theta_c}{\theta_i^o}$ is

$$\frac{\theta_c}{\theta_i^o} = 0.58 \Rightarrow 0.58 = \frac{T_0 - 25}{225 - 25}$$

$$\therefore T_0 = \underline{\underline{141^\circ C}}$$

2) Temp at depth of 10 mm from the surface (x/L)

$$x = L_c - \text{depth} \\ = 25 - 10 = 15 \text{ mm}$$

$$\therefore \frac{x}{L} = \frac{15}{25} = 0.6$$

from data book $Bi^0 = 0.21$ & $x/L = 0.6$

$$\frac{\theta}{\theta_c} = 0.95$$

$$\frac{\theta}{\theta_c} = \frac{T_{x/L} - T_0}{T_0 - T_0}$$

$$0.95 = \frac{T_{x/L} - 25}{141 - 25} \Rightarrow T_{x/L} = \underline{\underline{135.2^\circ\text{C}}}$$

3) Heat Loss from the plate during 2 sec

From data book Bi^0 & $Bi^{0^2} Fo$ charts

$$Bi^0 = 0.21$$

$$Bi^{0^2} Fo = 0.21^2 \times 3.07 \\ = 0.135$$

$$\therefore \frac{Q}{Q_i^0} = 0.45 \Rightarrow Q = Q_i^0 \times 0.45$$

$$\therefore Q_i^0 = m c_p dT$$

$$= \rho V c_p dT$$

$$= \rho (A \times L_c) \times c_p (T_f - T_0)$$

$$= 7850 \left(\frac{2 \times}{0.25} \right) \times 460 (225 - 25)$$

$$= 35.33 \times 10^3 \text{ kJ/m}^2$$

$$\therefore Q = 0.45 \times 35.33 \times 10^3$$

$$= \underline{\underline{15.9 \times 10^3 \text{ kJ/m}^2}}$$

2) The nose section of a missile is formed of a 6mm thick stainless steel plate and is held initially at uniform temp of 88°C . The missile enters the denser layers of the atmosphere at a very high velocity. The effective temp of air surrounding the nose section attains the value 2200°C , and the surface convective coefficient is estimated at $3405 \text{ W/m}^2\text{K}$. Make calculations for the max permissible time in these surroundings if the max metal temp is not to exceed 1095°C . Also work out the inside surface temp under these conditions. The constant values for steel properties are $\rho = 7850 \text{ kg/m}^3$, $c_p = 465 \text{ J/kgK}$, $k = 54 \text{ W/mK}$

Data

$$2L = 6\text{mm}, \Rightarrow L_c = 6/L = 3\text{mm} = 3 \times 10^{-3} \text{ m}$$

$$T_i = 88^\circ\text{C}, T_\infty = 2200^\circ\text{C}, h = 3405 \text{ W/m}^2\text{K}, T_0 = 1095^\circ\text{C} \text{ (Find } t \text{)}$$

$$\rho = 7800 \text{ kg/m}^3, c = 465 \text{ J/kgK}, k = 54 \text{ W/mK}$$

Find:-

1) $t = ?$ for reaching 1095°C at outside surface ($x = L_c$)

2) $T_0 = ?$ at inside surface ($x = 0$)

Assumptions :-

Sol

$$Bi^0 = \frac{hL_c}{k} = \frac{3405 \times 3 \times 10^{-3}}{54} = 0.189, Bi^0 > 0.1$$

$$Fo = \frac{\alpha t}{L_c^2} = \frac{k}{\rho c_p} \times \frac{t}{L_c^2}, \text{ we should find out 't'}$$

$$\frac{\theta_c}{\theta_i^0} = \frac{T_0 - T_\infty}{T_i^0 - T_\infty} = \frac{1095 - 2200}{88 - 2200} \approx 0.5232$$

1) From data book of Heisler charts of $\frac{\theta_c}{\theta_i^0} = 0.5232$ & $Bi^0 = 0.189$.

$$Fo = 3.6$$

$$\therefore Fo = \frac{k}{\rho c_p} \times \frac{t}{L_c^2} \Rightarrow 3.6 = \frac{54 \times t}{7800 \times 465 \times (3 \times 10^{-3})^2}$$

$$\therefore t = \underline{2.17 \text{ sec}}$$

2) Inside temp (take at $x=0$)

$$\frac{\theta_c}{\theta_i^0} = 0.5232$$

$$\frac{T_0 - T_\infty}{T_i^0 - T_\infty} \Rightarrow 0.5232 \Rightarrow \frac{T_0 - 2200}{88 - 2200} = 0.5232$$

$$\therefore T_0 = 1095^\circ\text{C}$$

3) A long ~~cylinder~~ Aluminium cylinder 5cm in diameter and initially at 200°C is suddenly exposed to convection environment at 70°C with heat transfer co-efficient of $525 \text{ W/m}^2\text{K}$. calculate the temp at the radius of 1.25cm 1 minute after the cylinder is exposed to the environment. Take $\rho = 2700 \text{ kg/m}^3$, $c = 900 \text{ J/kgK}$, $k = 210 \text{ W/m}^2\text{K}$.

Data

$$d_0 = 5 \text{ cm}, \quad r_0 = 2.5 \text{ cm} = 0.025 \text{ m}, \quad t = 1 \text{ min} = 60 \text{ sec}$$

$$T_i = 200^{\circ}\text{C}, \quad T_{\infty} = 70^{\circ}\text{C}, \quad h = 525 \text{ W/m}^2\text{K}$$

$$\rho = 2700 \text{ kg/m}^3$$

Find:-

$$1) T_{x/2} = ?$$

Φ

Assumptions:-

Sol

First find out centre line Temp from Heisler Chart

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_{\infty}}{T_i - T_{\infty}}$$

$$B_i = \frac{hL_c}{K} = \frac{525 \times 0.025}{210} = 0.0625$$

$$Fo = \frac{\alpha t}{L_c^2} \Rightarrow \frac{k}{\rho c} \times \frac{t}{L_c^2} \Rightarrow \frac{210 \times 60}{2700 \times 900 \times 0.025^2}$$

$$Fo = 8.29$$

From Heisler chart with Fo & Bi :

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_\infty}{T_i - T_\infty} = 0.2$$

$$\frac{T_0 - 70}{200 - 70} = 0.2 \Rightarrow T_0 = 96^\circ\text{C}$$

$$\text{Temp at any position } (r/r_0) = \frac{1.25}{2.5} = 0.5$$

$\therefore r/r_0$ & Bi charts

$$\frac{\theta}{\theta_c} = \frac{T(r/r_0) - T_\infty}{T_0 - T_\infty} \Rightarrow 0.78 = \frac{T(r/r_0) - 70}{96 - 70}$$

$$\therefore \underline{\underline{T(r/r_0) = 90.28^\circ\text{C}}}$$

- 4) A long cylinder shaft 20cm in diameter is made of steel ($k = 14.9 \text{ W/m}^\circ\text{K}$), $\rho = 7900 \text{ kg/m}^3$, $C = 477 \text{ J/kg}^\circ\text{K}$ and $\alpha = 3.95 \times 10^{-6} \text{ m}^2/\text{s}$. It comes out an oven at uniform temp of 600°C . The shaft is then allowed to cool slowly in an environment at 200°C with convective heat transfer co-efficient of $80 \text{ W/m}^2\text{K}$. Calculate the temp at the centre of the cylinder ~~after~~ 5-minutes after the start of cooling process. Also calculate the heat transferred per unit length of the shaft during this period.

Data

$$d = 20 \text{ cm}, \quad r_0 = 10 \text{ cm} = 0.1 \text{ m}, \quad k = 14.9 \text{ W/mK}$$

$$\rho = 7900 \text{ kg/m}^3, \quad c = 477 \text{ J/kgK}, \quad \alpha = 3.95 \times 10^{-6} \text{ m}^2/\text{s}$$

$$T_i = 600^\circ\text{C}, \quad T_\infty = 200^\circ\text{C}, \quad h = 80 \text{ W/m}^2\text{K}, \quad t = 45 \text{ min}$$

$$= 2700 \text{ sec.}$$

Find:-

- 1) $T_0 = ?$ at centre of the shaft after 45 min
- 2) Heat transfer (Q) = ?

Assumptions:-Sol:-

$$Bi = \frac{hr_0}{k} = \frac{80 \times 0.1}{14.9} = 0.537$$

$$Fo = \frac{\alpha t}{r_0^2} \Rightarrow Fo = \frac{3.95 \times 10^{-6} \times 2700}{(0.1)^2} = 1.0665$$

From data book centre temp charts with Bi & Fo

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_\infty}{T_i - T_\infty} = 0.5$$

$$\frac{T_0 - 200}{600 - 200} = 0.5 \Rightarrow T_0 = 400^\circ\text{C.}$$

From data book Heat-flux chart with Bi & $Bi^2 Fo$

$$Bi^2 Fo = (0.537)^2 \times 1.0665$$

$$= 0.3075$$

$$\therefore \frac{Q}{Q_0} = 0.62 \Rightarrow Q = 0.62 \times Q_0$$

$$Q_0 = m c_p dT \Rightarrow \int V c_p dT \Rightarrow \int \times \frac{\pi}{6} r_0^3 \times c_p (T_0 - T_\infty)$$

$$\therefore Q_0 = 0.62 \times 7900 \times \frac{\pi}{6} \times (0.1)^3 \times 1 \times 477 (600 - 200)$$

$$\therefore Q = \underline{\underline{29.36 \text{ MJ}}}$$

5) An egg can be approximated as a sphere, 5cm in dia with thermophysical properties $k = 0.6 \text{ W/mK}$, $\alpha = 0.14 \times 10^6 \text{ m}^2/\text{s}$. The egg is taken from a refrigerator at 2°C and is dropped into boiling water, where the convection heat transfer coefficient estimated as $1200 \text{ W/m}^2\text{K}$. Calculate the time required to reach the centre temp of the egg to 75°C .

Data

$$d = 5\text{cm}, \Rightarrow r_0 = 2.5\text{cm} = 0.025\text{m}, k = 0.6 \text{ W/mK}$$

$$\alpha = 0.14 \times 10^6 \text{ m}^2/\text{s}, T_i = 2^\circ\text{C}, T_\infty = 100^\circ\text{C} (\text{Dropped in Boiling water}), h = 1200 \text{ W/m}^2\text{K}, T_0 = 75^\circ\text{C}.$$

Find:-

1) $t = ?$ for reaching 75°C

Assumptions:-

Sol

$$B_i = \frac{h r_0}{k} = \frac{1200 \times 0.025}{0.6} = 50.$$

$$Fo = \frac{\alpha t}{r_0^2} \Rightarrow \text{Here we should find out } t$$

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_\infty}{T_i - T_\infty} = \frac{75 - 100}{2 - 100} = 0.255$$

with $\frac{\theta_c}{\theta_i}$ & B_i .

we donot have any chart with $B_i = 50$
(one term solution)

so we can use ~~high~~ B_i numbers i.e.,

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_\infty}{T_i - T_\infty} = C_1 e^{-\xi^2 Fo}$$

↳ one term solution

with sphere $B_i = 50$ from tables

$$\xi = 3.0788 \text{ \& } C = 1.9962$$

$$\therefore \frac{75 - 100}{2 - 100} = 1.9962 e^{-(3.0788)^2 \times Fo}$$

$$\therefore Fo = 0.2154 \Rightarrow Fo = \frac{\alpha t}{L_c^2}$$

$$0.2154 = \frac{0.14 \times 10^{-6} \times t}{(0.025)^2}$$

$$\therefore t = 961.66 \text{ sec} \Rightarrow t = \underline{\underline{16 \text{ min}}}$$

6) A 3.6 cm diameter egg, approximated spherical in shape, is initially at 25°C . Temp to boil it to the consumer's taste, it needs to be placed 225 seconds in a saucepan of boiling water at 100°C . For how long should a similar egg for the same consumer be boiled if egg taken from a refrigerator at temp 5°C . Thermophysical properties of egg is $k = 2.5 \text{ W/mK}$, $\rho = 1250 \text{ kg/m}^3$, $c = 2200 \text{ J/kgK}$, $h = 2800 \text{ W/m}^2\text{K}$.

Data

$$d = 3.6 \text{ cm}, \quad r_0 = 0.18 \text{ cm} = 0.18 \times 10^{-2} \text{ m}$$

$$T_i = 25^\circ\text{C}, \quad T_\infty = 100^\circ\text{C}, \quad t = 225 \text{ sec.}$$

For similar taken from refrigerator

$$T_i = 5^\circ\text{C}, \quad T_\infty = 100^\circ\text{C}.$$

$$k = 2.5$$

Find:-

- 1) For time $t = 225 \text{ sec}$ $T_0 = ?$ at $T_i = 25^\circ\text{C}$
- 2) For T_0 time $t = ?$ at $T_i = 5^\circ\text{C}$.

Assumptions:-

Sol

$$Bi = \frac{h r_0}{k} = \frac{2800 \times 0.18 \times 10^{-2}}{2.5} = Bi$$

$$\therefore Bi = 2.016$$

$$Fo = \frac{\alpha t}{L_c^2} \Rightarrow Fo = \frac{K}{\rho c_p} \frac{t}{L_c^2}$$

$$Fo = \frac{2.5 \times 225}{1250 \times 2200 \times (0.18 \times 10^{-2})^2}$$

$$Fo = 63.13$$

From data book centre line temp [because in ^{sphere} ~~uniform~~ the HT towards centre] of Heisler charts with Bi & Fo

$$\frac{\theta_c}{\theta_i} = \frac{T_0 - T_\infty}{T_i - T_\infty} = 0.2$$

$$\frac{T_0 - 100}{25 - 100} = 0.2$$

$$\therefore T_0 = \underline{\underline{83.5^\circ C}}$$

For similar egg at $5^\circ C$ to reach $T_0 =$

$$\frac{T_0 - T_\infty}{T_i - T_\infty} \Rightarrow \frac{83.5 - 100}{5 - 100} = 0.174$$

$\therefore$ from $\frac{\theta_c}{\theta_i}$ & Bi & Im Heisler chart

$$Fo =$$

$$Fo = 0.71$$

$$\therefore F_0 = \frac{\alpha t}{L_c^2}$$

$$= \frac{k}{\rho c_p} \frac{t}{L_c^2}$$

$$= \frac{2.5 \times t}{1250 \times 2200 \times (0.18 \times 10^{-2})^2}$$

$$\therefore t = \underline{\underline{2.53 \text{ seconds}}}$$